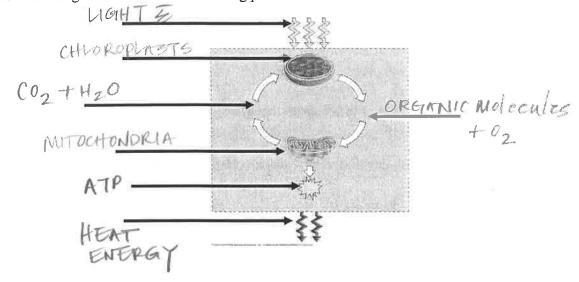
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Name Period Chapter 9: Cellular Respiration: Harvesting Chemical Energy

Overview: Before getting involved with the details of cellular respiration and photosynthesis, take a second to look at the big picture. Photosynthesis and cellular respiration are key ecological concepts involved with energy flow. Use Figure 9.2 to label the missing parts below.



Concept 9.1 Catabolic pathways yield energy by oxidizing organic fuels

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CU	icopi 7.11 Cuiubone punitruys yiona onongy of stranging organic filess
1	Explain the difference between fermentation and cellular respiration.
	NO O2 (ANAGROBIC) REQUIRES 02 IN MITOCHONDRIA
2.	Give the formula (with names) for the catabolic degradation of glucose by cellular respiration.
	Both cellular respiration and photosynthesis are redox reactions. In redox, reactions pay attention to
3.	Both cellular respiration and photosynthesis are redox reactions. In redox, reactions pay attention to
	the flow of electrons. What is the difference between oxidation and reduction?
	OXIDATION = AGENT LOSES &
	PEDUCTION = ABENT GAINS &
4.	
	$Xe^- + Y \rightarrow X + Ye^{\text{REDUCEO}}$
	$Xe^- + Y \rightarrow X + Ye^-$

Draw an arrow showing which part of the reaction is oxidized and which part is reduced.

is the reducing agent in this reaction, and is the oxidizing agent.

- When compounds lose electrons, they LOSE energy; when compounds gain electrons, they GAIN energy.
- In cellular respiration, electrons are not transferred directly from glucose to oxygen. Each electron is coupled with a proton to form a hydrogen atom. Following the movement of hydrogens allows you to follow the flow of electrons. The hydrogens are held in the cell temporarily by what electron carrier?

NAD+, A COENTYME
What is a coenzyme? (If you have forgotten, look back to a few pages in Chapter 8.)

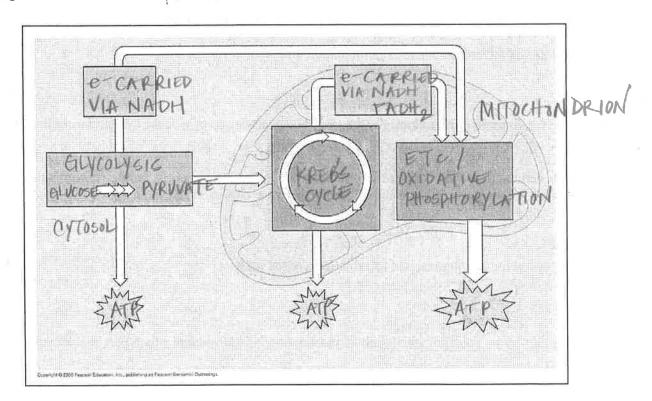
What is the function of the electron transport chain in cellular respiration?

BREAKS THE FALL OF E- TO DXYGEN INTO SEVERAL ENERGY

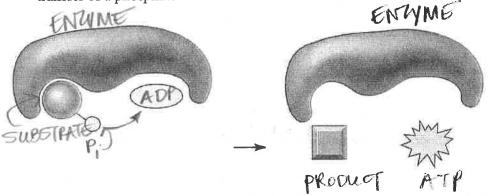
Show the normal, downhill route most electrons follow in cellular respiration: RELEASING STOPS

Glucose > NADH -> ELECTRON -> oxygen
TRANSPORT CHAIN

9. Understanding the overall map of how cellular respiration works will make the details easier to learn. Use Figure 9.2 to label the missing information in the figure below.



- 10. Three types of phosphorylation (adding a phosphate) are covered in the text, and two of these occur in cellular respiration. Explain how the electron transport chain is utilized in *oxidative phosphorylation*.
- 11. The second form of phosphorylation is *substrate level*. Label the figure below to show the direct transfer of a phosphate from a substrate to ADP to form ATP.



Concept 9.2 Glycolysis harvests chemical energy by oxidizing glucose to pyruvate

12. Why is glycolysis an appropriate term for this step of cellular respiration?

GLUCOSE 15 SPLT (INTO 2 3-C DYKWATES)

14. The ten individual steps of glycolysis can be divided into two stages: energy investment and energy payoff. Label the energy investment stage below; then use Figure 9.9 to find the two specific stages where ATP is used.

NERGY GLUCOSE
INVESTMENT 2ADP WEATP

15. The second step in glycolysis is the *energy payoff phase*. Label this stage. Note that it provides both ATP and NADH. Look at Figure 9.9 to locate the two stages where ATP is formed and the one stage where NADH is formed.

PAYDEF

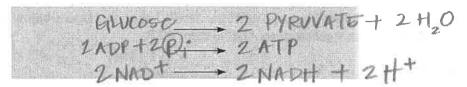
HADPHHILL
STAGE 10

TO STAGE 10

TO

STAGE 2

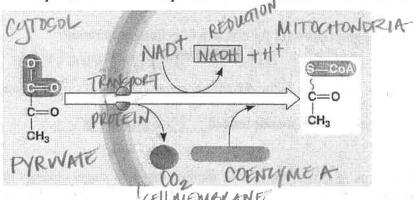
16. This final figure shows the net gain of energy for the cell after glycolysis. Most of the energy is still present in the two molecules of pyruvate. Fill in the chart below and show the net energy gains.



17. Notice that glycolysis occurs in the _______ of the cell. What is the relationship concerning glycolysis and oxygen?

IT OCCURS IN THE ABSENCE OF O_2 Concept 9.3 The citric acid cycle completes the energy-yielding oxidation of organic molecules

18. To enter the citric acid cycle, pyruvate must enter the mitochondria by active transport. Three things are necessary to convert pyruvate to acetyl CoA. Complete the missing parts of the chart below and then explain the three steps in the conversion process.



(1)
PYRWATE'S CARBOXYLGIROUP(-COO) IS REMOVED & GIVEN OFF AS A MOLECULE OF CO.

(2) REMAINING & CARBON FRAGMENT IS DXIDIZED, FORMING ACETATES. ENTYME TRANSFERS & TO NADY STORING & AS NADH.

(3) COA (COENCYMEA) IS ATTACHED TO ACETATE BY AN UNSTABLE BOND THAT MAKES THE ACETYLAROUP REACTIVE

19. How many times does the citric acid cycle occur for each molecule of glucose?

20.	Use Figure 9.11 to help you answer the following summary questions about the citric acid cycle:
	a. How many NADHs are formed? 3
	b. How many total carbons are lost as pyruvate is oxidized?
	c. The carbons have been lost in the molecule CARBON DIOXI.DE
	d. How many FADH ₂ have been formed?
	e. How many ATPs are formed?
21.	The diagram covers only one pyruvate, although two pyruvates are formed from a single glucose. How many molecules of the following are formed from the breakdown of glucose?
	a. NADH =
	b. $FADH_2 = $
	c. ATP = \bigcirc
22.	The step that converts pyruvate to acetyl CoA at the top of the diagram also occurs twice per glucose.
	This step accounts for two additional reduced NAPH molecules and two carbon dioxide
	molecules.
23.	Explain what has happened to the six-carbon molecules found in the original glucose molecule.
	RELEASED AS COD; ONLY 2 ATTS PRODUCED
	ncept 9.4 During oxidative phosphorylation, chemiosmosis couples electron transport to ATP thesis
24.	Oxidative phosphorylation involves two components: the electron transport chain and ATP synthesis.
	Referring to Figure 9.13, notice that each member of the electron transport chain is lower in free
	ENERGY than the preceding member of the chain, but higher in Electrone the molecule
	at zero free energy, which is
	electronegativity.

25. Explain why oxygen is the ultimate electron acceptor. Oxygen stabilizes the electrons by combining with two hydrogen ions to form what compound? > WATER (H20)

IT ISVERY ELECTRON EGAMINE 26. The two electron carrier molecules that feed electrons into the electron transport system are

NADH and FADH

27. Using Figure 9.14, explain the overall concept of how ATP synthase uses the flow of hydrogen ions to produce ATP.

1. HT IONS FLOWING DOWN GRADIENT ENTER A & CHANNEL IN A STATOF-ANGIORED IN MEMBRANE. 2. HT ENTER BINDING SITES WITH A ROTOR-CHANGING SHAPE OF SUBVINITY CO LITER SPINS WI MEMBRANE.

3. EACH HIT MAKES COMPLETE THEN, LEAVES TO PASSIFIE A 21 & CHANNEL 28. What is the role of the electron transport chain in forming the HT gradient across the inner 4. Spinling

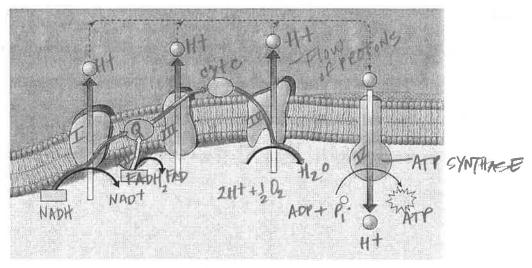
mitochondrial membrane?

Z CONVERTER USING EXERGONIC FLOW OF C-FROM NATOH -> FATH TO pump tit ACROSS MEMBEANE

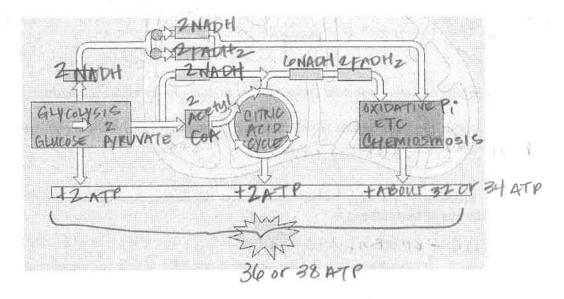
PLOM MITOCATED DE AL MATRIX TO INTERMEMBERGE SPACE.
29. Two key terms are chemiosmosis and proton-motive force. Relate both of these terms to the process of oxidative phosphorylation.

CHEMIOSMOSIS AND THE ETC COMPOSE OXIDATIVE PHOSPITORYLATION ...

30. Figure 9.16 is a key to understanding the production of ATP in the mitochondria. In the figure below, label all locations and molecules. Then use one color to trace the flow of electrons and another color to show the flow of protons.



- 31. At this point, you should be able to account for the total number of ATPs that could be formed from a glucose molecule. To accomplish this, we have to add the substrate-level ATPs from glycolysis and the citric acid cycle to the ATPs formed by chemiosmosis. Each NADH can form a maximum of ATP molecules. Each FADH₂, which donates electrons that activate only two proton pumps, makes ATP molecules.
- 32. Use the figure to account for all the ATP molecules formed during cellular respiration. Use the text to be sure you understand how each subtotal on the bar below the figure is reached.



35. For aerobic respiration to continue, the cell must be supplied with oxygen—the ultimate electron acceptor. What is the electron acceptor in fermentation?

AN ORGANIC MOLECULE SUCH AS PYRVVATE (LACTIC ACID)

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- 36. Explain how alcohol fermentation starts with glucose and yields ethanol. Be sure to stress how NAD⁺ is recycled. RELEASED CO_ FROM PYRVVATE IS CONVERTED -> ACETAL DEHYDE WHICH IS REDUCED by NADH > ETHANOL.
- 37. Explain how lactic acid fermentation starts with glucose and yields lactate. Be sure to stress how NAD is recycled. WOUNDERS DEFINITED DID FAT Y BY NADILL TO

TYRVATE (S KOULDS DIRECTLY BY NADE) TO FORM
LACTATE AS AN END PRODUCT WITH NO RELEASE
0FC02.
38. Using Figure 9.19 as a guide, draw and explain why pyruvate is a key juncture in metabolism.
MITOCHONDRION WHAT HAPPEN
ETHANOL PYRUVATE PRESENT ACCIVICAN LACTATE Concept 9.6 Glycolysis and the citric acid cycle connect to many other metabolic pathways OF NOT
39. What three organic macromolecules are often utilized to make ATP by cellular respiration? PROTEING, CARBOHY DRATES, FATS
40. Explain the difference in energy usage between the catabolic reactions of cellular respiration and anabolic pathways of biosynthesis.
CATROOLIC - ENERGY ISGENERATED; BREAKDOWN MOLECULES
ANABOLIC - ENERGY CONSUMED TO BUILD UP MOLECULES
41. Explain how AMP stimulates cellular respiration while citrate and ATP inhibit it.
AMP (PRODUCT OF ATP) ACCUMULATION IS ASIGNAUTITAT MORE
ATP BUILD UP INTIBITS THE ENVINE, SLOWING DOWN GLYCOLYSIS. Testing Your Knowledge: Self-Quiz Answers Now you should be ready to test your knowledge. Place your answers here:
123456789